



Study of effects of Groundnut shell and Grog additives on the refractory Properties of Isiagu Clay

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Abstract: This research aimed at finding out the impacts of groundnut shell and grog additives on the refractory properties of Isiagu Clay. The raw materials were collected locally, processed and analysed using Scanning Electron Microscope/Energy Dispersive X-ray (SEM/EDX). The additives were added to the raw clay in the ratio of 2.5, 5.0, 7.5 and 10 wt% and fired at 900°C and 1100°C, respectively. The refractory properties measured were; linear shrinkage, apparent porosity and bulk density. The result of the SEM/EDX analyses showed that the clay is a fire clay since it contains 61.68% Alumina (Al_2O_3) and 34.97% Silica (SiO_2) while the grog additive contains 65.5% Al_2O_3 and 34.33% SiO_2 , respectively. The groundnut shell itself contains 1.55% Al_2O_3 and 2.91% SiO_2 , respectively. Groundnut shell also contains 75.1% Nb_2O_5 which imparts good heat conductivity to samples containing it. SEM/EDX morphological analysis results showed that both grog and groundnut shells contain tiny to coarse particles, however, grog has greater strength for refractory application than groundnut shell due to its high alumina and silica contents. For groundnut shells, linear shrinkage increased from 7.477% to 7.881% at 900°C while grog decreased from 3.383 to 3.509 % at 900°C. The decrement in linear shrinkage in grog-containing samples is due to the high amount of Al_2O_3 and SiO_2 present in the grog which makes it desirable for refractory application. The apparent porosity measurement showed that the samples containing groundnut shell are suitable for insulating application since it contains a high amount of organic matter. ANOVA results confirmed that temperature and additive percentages were not significant for linear shrinkage measurement.

Keywords: Groundnut shell, grog, Clay, fireclay, insulating, and refractory

1. Introduction

Heat transfer is paramount in the process industry of which heating, boiling and roasting of metal ores are classical examples [1]. In these processes, the aim is to achieve efficient heat transfer in process equipment such as ovens, furnaces and boilers by lining the interior with high-temperature materials [2]. In Nigeria, seven clay deposits exist which are of good refractory properties and hence good feedstock for the production of high-temperature bricks at scale [3, 4]. According to Olalere, *et al* [5] and Oke, *et al* [4], the Nigerian government is set to rehabilitate her obsolete steel industries in the Kogi and Delta states which would require 68,000 tons of refractories which is only a small fraction of the refractory demands of other energy

industries totalling 300,000 tons in the year 2000. Sadly, about \$229 million is spent annually to cushion the unavailability of refractory materials for use in kilns and furnaces [6]. Although, some efforts have been made to meet this need, yet, there is a need to provide these high-temperature materials with improved mechanical and thermal properties. This must be done since weakness is usually due to chemical attacks and mechanical abrasions in the process of utilizing them, hence decreasing their life span to as low as 1 to 2 years [7]. Local clays can be used together with additives to improve the various properties of the refractory clay-like; thermal shock resistance, cold crushing strength, bulk density, porosity etc [8, 9]. Some researchers have made efforts to improve the quality of high-temperature

materials for furnaces, boilers and oven linings. Improvement of thermal performance of environmental friendly bricks was studied by Kazmi, *et al* [10] in which they used agro-waste sugarcane bagasse ash (SBA) and Rice husk ash (RHA) as additives. This additive was incorporated in various dosages (5%, 10% and 15%) by the weight of clay. After various tests were carried out it was discovered that refractory brick incorporating agro-waste had lightweight and low compressive strength. In this present study, grog from clay and groundnut shell were used to improve the refractory properties of Isiagu clay. The Analysis of Variance (ANOVA) was achieved using Excel 2013 and the degree of significance of the factors was determined.

2. Materials and Methods

2.1 Materials

The raw clay sample was air-dried under shade to allow removal of water and other volatile matter and afterwards crushed to a small grain size in a mortar to increase the surface area of the clay sample [3, 11]. To get rid of unwanted materials, the slurry was passed through a 0.425mm mesh sieve [11]. The clay slip obtained was sun-dried for two days and then oven-dried at 100°C and fired at 600°C to remove carbonate and other organic matter [11]. Afterwards, some reasonable part of the clay was fired up to 1000°C for 6 hours to produce grog which was an additive just as did [12]. The processed clay was pulverized and sieved through a 425µm sieve which is recommended by ASTM [13]. The clay sample was mixed proportionally with the additives and moulded into different shapes namely; rectangular, circular, and conical for the respective tests to be carried out. The morphological analysis was carried out using the Zeiss EVO LS 10. A beam of electrons is produced at the top of the microscope using an electron gun. The beam was focused toward the sample via the magnetic field which ejected electrons and x-rays as they contacted. The detector read the signals and converted them to the final image.

2.2 Mixing and Molding

The clay sample was combined individually with the additives namely; grog, groundnut shell, melon shell, palm kernel shell, sawdust and chicken droppings (2.5%wt, 5.0wt%, 7.5wt%, and 10%wt). Following the addition of water, a sticky mass was formed and with the aid of the mould, the samples were formed for different tests to be conducted [11]. The moulds were rubbed with lubricating oil for easy release of the test pieces from the mould after it dries [11]. The formulation of the samples for analysis is shown in Table 1.

Table 1. Design of experiment for the production of refractory

Input variables	Ranges
Temperature (°C)	900 – 1100
Additive percentage (wt%)	2.5 – 10

2.3 Measurement and Analysis

a. Linear Shrinkage

This evaluates the changes in the length of brick samples when heated. For this purpose, the brick samples will be made in cuboidal shapes. The formula is as given in the work of Adeosun, *et al* [14]. This is calculated via the original length (L_o) before drying and the final length (L_f) after firing to a certain temperature

$$\text{Percentage Total Shrinkage} = \frac{L_o - L_f}{L_o} \times 100 \quad (2.1)$$

b. Apparent Porosity

Apparent porosity is the mathematical relationship between the volume of the void and the total volume of the material as given by ASTM-C20-00 [15]. This will be determined using the boiling method as was used by Adeniyi, *et al* [16]. A moulded brick specimen (rectangular) will be used. The brick will be oven-dried at 110°C to constant weight (D). After which it will be transferred to a beaker and boiled with distilled water for 1.5hrs to assist in releasing the trapped air. It will be soaked and the saturated weight free of water (W) will be obtained. Finally, the specimen will be suspended in the water by a rope tied to a spring balance and obtain the suspended weight (S) when it is completely immersed in water.

$$\text{Apparent Porosity} = \%Pa = \frac{W-D}{W-S} \times 100 \quad (2.2)$$

Where:

D = Constant Weight of the dry Sample

S= Weight of the sample suspended in the water

W= Weight of sample in the air including the moisture in its open pores (saturated weight)

c. Bulk Density

The Bulk Density (BD) of a refractory shows the degree to which it was fired and hence the level of densification will be determined by dividing the mass of sample by the outside volume and multiplying with the density of water as recommended by ASTM-C20-00 [15].

$$\text{Bulk Density} = \frac{\text{Dry weight of sample} \times \text{Density of water}}{\text{saturated weight in air} - \text{suspended weight in water}} \quad (2.3)$$

d. Refractoriness

This is a thermal property of refractory brick materials that determines the degree of temperature the material can withstand. This will be achieved by making conical brick samples and putting them alongside pyrometric sega cones (standard cones with definite deformation temperature) into the kiln and firing to a very high temperature of about 1650°C. After firing, the cones will be examined and the one that bent to the same extent as the test sample is said to have the equivalent temperature of deformation. Alternative, Shuen's formula can be used to roughly estimate the refractoriness of a clay sample [8].

$$\text{Refractoriness (K)} = \frac{360 + Al_2O_3 - RO}{0.228} \quad (2.4)$$

Where RO is all oxides minus Al₂O₃ and SiO₂

RESULTS AND DISCUSSION

3.1 Characterization of Clay and Additives

The results obtained from the characterization of clay and additive samples using SEM/EDX was carried out at an Electron High Tension of 10kV at different wavelengths of 100µm and 20µm and the result are shown in Table 2. SEM/EDX results for the weight composition of elements show that the clay contains principally, aluminium and silicon elements with a relatively high amount of carbon [17]. The result presented in Table 2 shows that the sawdust additive contains only carbon metal, this is since sawdust is very rich in organic matter compared with coconut shell and rice husk [18]. Grog on the other hand is very high in silicon as expected because it was gotten from a clay sample fired to 900°C where all the carbonaceous materials are removed [12]. The result in Table 3 below was derived from the results in Table 2. The oxide forms of the elements present were obtained and the result showed that clay and grog had almost the same

composition by weight of both Al₂O₃ and SiO₂. Since both samples contain a high amount of Al₂O₃ and SiO₂, they can be referred to as kaolinites, hence suitable for the production of refractory bricks (Chikwelu, et al, 2018; Ajala & Badarulzaman, 2016a).

Table 2: SEM/EDX results for the percentage by weight composition of elements in clay and all additives

Weight composition	MgO	Al ₂ O ₃	P ₂ O ₅	SiO ₂	K ₂ O	Nb ₂ O ₅
Clay	0.27	34.97		61.68	3.08	
Grog	0.17	34.33		65.5		
Groundnut shell		5.40		5.75	13.15	75.71

Table 3: SEM/EDX results for the percentage by weight composition of oxides in the clay and all additives

Weight composition	Mg	Al	Si	K	Nb	C	O	N
Clay	0.3	17.37	54.01	2.4		24.44	1.47	
Grog	0.25	22.68	76.28				0.78	
Groundnut shell		1.55	2.91	5.92	28.11	50.39	1.64	9.49

3.2 SEM/EDX Morphological analysis result for Clay, groundnut and grog

SEM/EDX morphological analysis results for various samples are presented in fig 2. The analysis was carried out at an Electron High Tension of 10kV at different wavelengths of 100µm and 20µm, respectively. The SEM/EDX result for the clay and the grog samples indicates the presence of tiny to large coarse particles (fig2a & b), however, Aboutaleb, et al [19] obtained the same result but with refractory brick waste (RBW). Also, the groundnut shell sample showed the presence of tiny to large coarse particles in form of a sheet l (fig 2c).

3.3 Refractory Analysis Results

a. Linear Shrinkage

The results of the linear shrinkage measurement are shown in Fig. 3 below. The results show that while grog additive decreased progressively, groundnut shell increased for both 900°C and 1100°C and was within the acceptable limit of 7 to 10% [11].

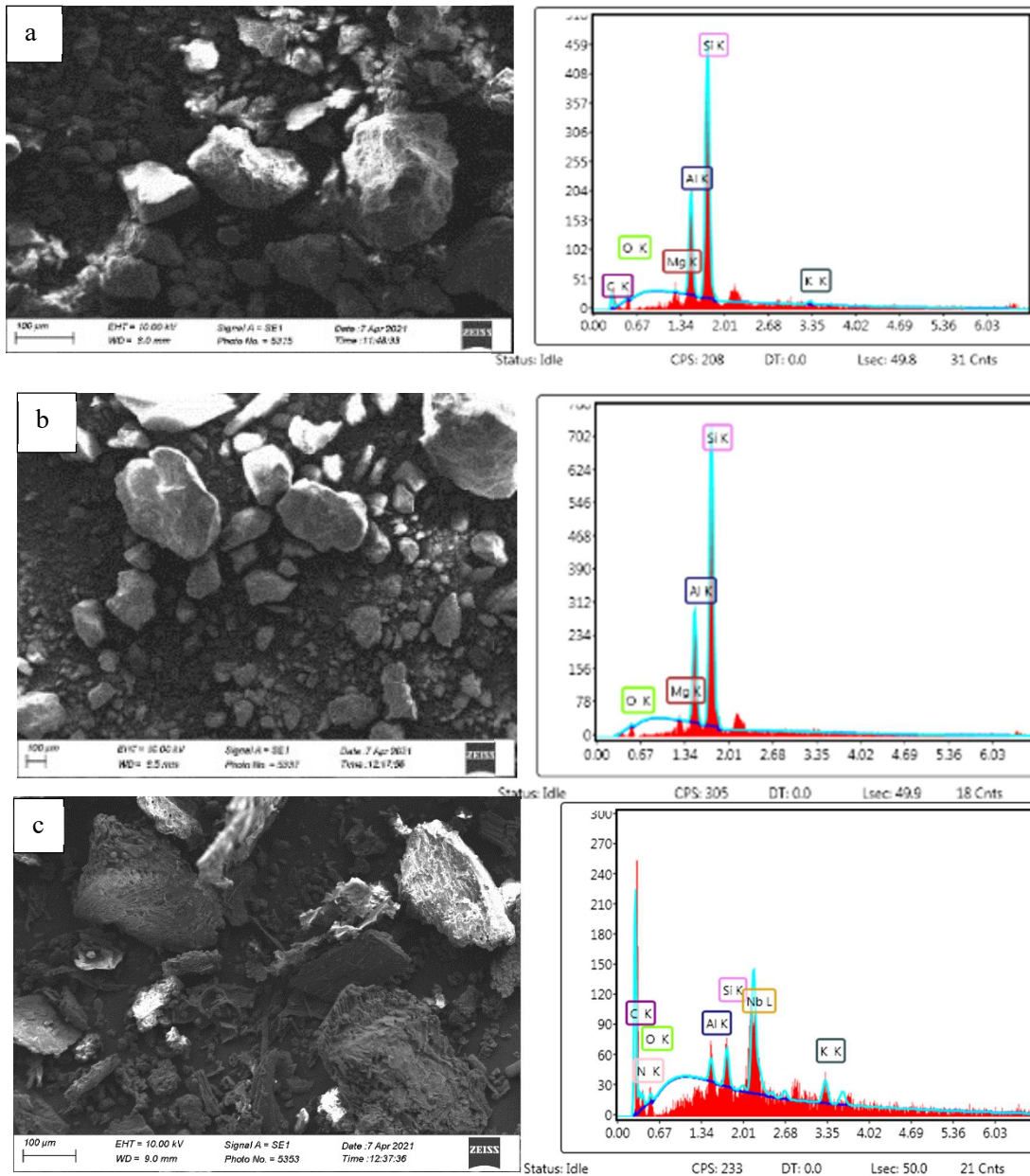
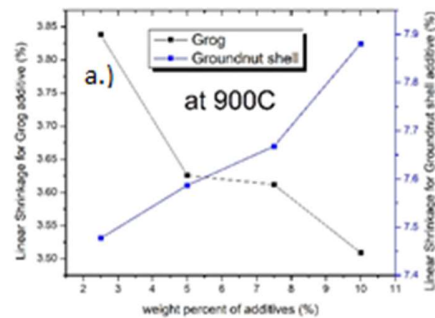


Fig 2. SEM/EDX results at EHT = 10.00kv and wavelength of 100 μ m for a.) Isiagu Clay sample b.) Grog and c.) Groundnut Shell

However, the variation of grog was the opposite of this trend and the least among the samples. This agrees with the SEM/EDX results which showed that grog contains more silicon than any other element. The result is shown in Fig 3. shows that grog and groundnut shells have comparable linear shrinkage at both 900oC and 11000C, that is, opposite trends were recorded for the additive percentage.



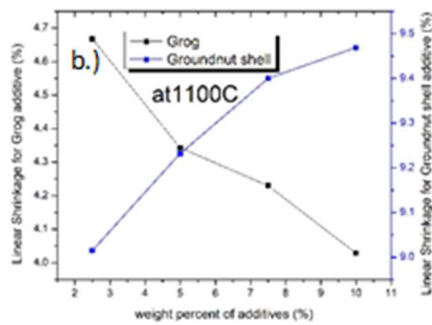


Fig 3. The plot of Linear Shrinkage for Groundnut shell and Grog at a.) 900°C and b.) 1100°C

a. Apparent Porosity

The apparent porosity increased steadily with the addition of groundnut shell additive but decreased with grog additives at both 900°C and 1100°C (Fig 4) [20]. Based on the international standard of 2 to 30%, the apparent porosity of the groundnut shell was within the specified limit [11].

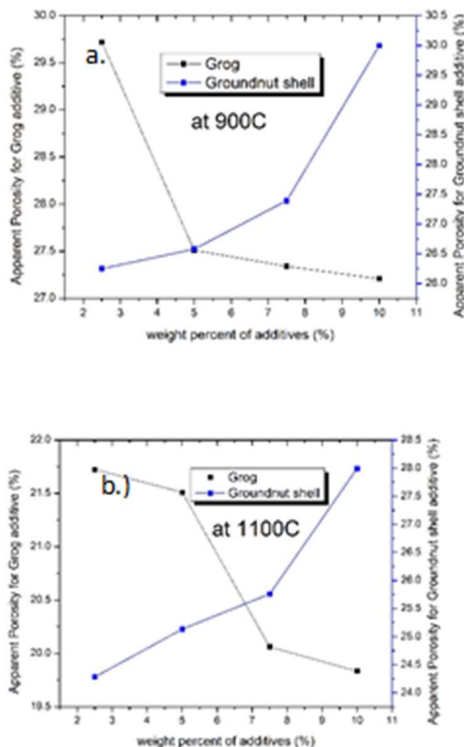


Fig 4. The plot of Apparent porosity for Sawdust and Grog at a.) 900°C and b.) 1100°C

b. Bulk Density

There was a bulk density decrement according to Fig. 5 following previous works [11, 21, 22]. A handful of

the samples containing grog at 1100°C were within the internationally acceptable range of 1.7 – 2.1 g/cm³ while the samples containing groundnut shell at 900°C and 1100°C were within the specified standard [11].

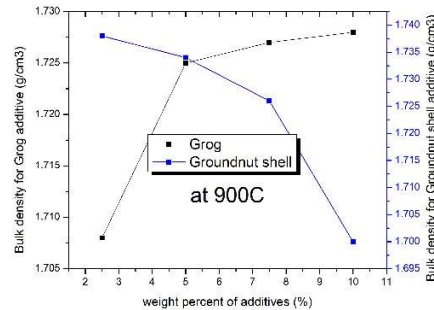


Fig 5. The plot of Bulk Density for Groundnut shell and Grog at a.) 900°C and b.) 1100°C

d. Refractoriness

This was determined using equation 2.4. The calculation indicated that Isiagu clay is fireclay and hence can withstand temperatures as high as 1447oC. This is very close to that recorded in the work of Harith and Hani [23] where a sintering temperature of 1500oC was observed for 70% kaolin and 30% metakaolin and above 1325oC recorded by Aşkın, et al [24] for Cordierite with waste magnesite.

3.4 ANOVA

Results for Grog and Sawdust Additives
Analysis of variance was done for all the samples with various additives and all the properties measured.

Grog Only additive

Analysis of variance was carried out for all samples containing grog additives. The result of two-way ANOVA without replication is presented in Table 4 to Table 6. The result shows that the effect of percentage additive for all properties was not significant ($P > 0.05$),

while the temperature was not significant only for linear shrinkage measurement [25, 26]. This is because organic matters have been removed from the grog by firing and hence increase in grog per cent had no impact on the properties measured.

Groundnut shell only

Analysis of variance was carried out for all samples containing groundnut shell additives. The result of two-way ANOVA without replication is presented in Table 7 to Table 9. The result shows that the effect of temperature and percentage additive for all properties were significant ($P < 0.05$), except for the effect of percentage additive for bulk density [25, 26].

Table 4: Linear Shrinkage

ANOVA								
Source of Variation	SS	df	MS	F	P-value	F crit	Remark	
Temperature	14.61114	2	7.305569	1.86858	0.233968	5.143253	Not significant	
Per cent Additive	8.061763	3	2.687254	0.687332	0.591915	4.757063	Not significant	
Error	23.45814	6	3.90969					
Total	46.13104	11						

Table 5: Apparent Porosity

ANOVA								
Source of Variation	SS	df	MS	F	P-value	F crit	Remark	
Temperature	967.1841045	2	483.6	85.06908	3.95E-05	5.143253	Significant	
Percentage Additives	2.52790825	3	0.843	0.148229	0.927107	4.757063	Not Significant	
Error	34.1081895	6	5.685					
Total	1003.820202	11						

Table 6: Bulk Density

ANOVA								
Source of Variation	SS	df	MS	F	P-value	F crit	Remark	
Temperature	53.83901	2	26.9195	7.794196	0.021468	5.14325285	Significant	
Per cent Additive	10.52781	3	3.509272	1.016064	0.448761	4.757062663	Not Significant	
Error	20.72273	6	3.453788					
Total	85.08955	11						

This implies that the firing had little or no impact on the weight of the samples from 2.5wt% to 10wt.

Table 7: Linear Shrinkage

Source of Variation	SS	df	MS	F	P-value	F crit	Remark
Temperature	207.5613	2	103.7807	68.81258	7.29E-05	5.14325285	Significant
Per cent Additive	30.4618	3	10.15393	6.732645	0.023907	4.757062663	Significant
Error	9.048983	6	1.508164				
Total	247.0721	11					

Table 8: Apparent Porosity

Source of Variation	SS	Df	MS	F	P-value	F crit	Remark
Temperature	1118.426	2	559.2128	519.3603	1.89E-07	5.143253	Significant
Per cent Additive	40.99877	3	13.66626	12.69232	0.005228	4.757063	Significant
Error	6.460403	6	1.076734				
Total	1165.885	11					

Table 9: Bulk Density

Source of Variation	SS	df	MS	F	P-value	F crit	Remark
Temperature	54.40336	2	27.20168	7.759389	0.021677	5.143253	Significant
Per cent Additive	10.21775	3	3.405917	0.971551	0.465523	4.757063	Not Significant
Error	21.03388	6	3.505647				
Total	85.65499	11					

4. CONCLUSION

The effect of grog and groundnut shells on the refractory properties of Isiagu clay was studied. The ANOVA was also carried out using Excel 2013. The SEM/EDX result showed that groundnut shell additive contains 50% carbon metal and implies that samples containing groundnut shell additive can be used for insulating firebrick production. In the same vein, the Grog additive showed high content of Al₂O₃ and SiO₂ which makes it suitable for refractory applications. The SEM/EDX morphological results further confirmed the class of application of the two additives used in this work. Grog and groundnut shell both contains tiny to coarse particles which indicates their strength, however, grog has an advantage over groundnut shell because of the

higher Al₂O₃ and SiO₂ content. Samples containing grog showed a reverse trend in all the properties measured due to the high content of Al₂O₃ and SiO₂ and further confirming its suitability for a refractory application. ANOVA results confirmed that for grog and groundnut shell additives, the additive percentage was not significant for bulk density which implies that the bulk density is Acknowledgement

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